Application No.: 10/525,818 Docket No.: 0171-1184PUS1

## AMENDMENTS TO THE CLAIMS

 (Previously presented) A nonaqueous electrolyte characterized by containing: an ionic liquid having general formula (2) below and a melting point not higher than 50°C

$$\begin{bmatrix} Me \\ Et - X - CH_2CH_2OR' \\ Et \end{bmatrix}^+ \cdot Y$$
 (2)

wherein R' is methyl or ethyl,

X is a nitrogen atom or a phosphorus atom,

Y is a monovalent anion,

Me stands for methyl and Et stands for ethyl;

a compound which reductively decomposes at a more noble potential than the ionic liquid; and

a lithium salt.

- 2. (Original) The nonaqueous electrolyte of claim 1 which is characterized in that the compound reductively decomposes at a more noble potential than the ionic liquid when a working electrode used with the electrolyte is made of a carbonaceous material or metallic lithium.
- 3. (Original) The nonaqueous electrolyte of claim 1 or 2 which is characterized in that the content of said compound within the electrolyte is from 0.1 to 60 wt%.
- 4. (Original) The nonaqueous electrolyte of claim 3 which is characterized in that the content of said compound is 0.1 to 30 wt%.

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5. (Previously presented) The nonaqueous electrolyte of claim 1 which is characterized in that the compound is one or more selected from among ethylene carbonate, propylene carbonate, vinylene carbonate, dimethyl carbonate, ethyl methyl carbonate and diethyl carbonate.

- 6. (Previously presented) The nonaqueous electrolyte of claim 1 which is characterized in that the ionic liquid has a melting point not higher than 25°C.
- 7. (Currently amended) The nonaqueous electrolyte of claim 1 which is characterized in that X is a nitrogen atom, nitrogen atom and R' is methyl, and the letter n is 2.
- 8. (Canceled)
- 9. (Previously presented) The nonaqueous electrolyte of claim 1 which is characterized in that Y is BF<sub>4</sub>, PF<sub>6</sub>, (CF<sub>3</sub>SO<sub>2</sub>)<sub>2</sub>N<sup>-</sup>, CF<sub>3</sub>SO<sub>3</sub> or CF<sub>3</sub>CO<sub>2</sub>.
- 10. (Previously presented) The nonaqueous electrolyte of claim 1 which is characterized in that the lithium salt is LiBF<sub>4</sub>, LiPF<sub>6</sub>, Li(CF<sub>3</sub>SO<sub>2</sub>)<sub>2</sub>N, LiCF<sub>3</sub>SO<sub>3</sub> or LiCF<sub>3</sub>CO<sub>2</sub>.
- 11. (Previously presented) A nonaqueous electrolyte secondary cell having a positive electrode which contains a lithium-containing double oxide, a negative electrode which contains a carbonaceous material capable of inserting and extracting lithium ions or contains metallic lithium, a separator between the positive and negative electrodes, and a nonaqueous electrolyte;

which secondary cell is characterized in that the nonaqueous electrolyte is a nonaqueous electrolyte according claim 1.

## 12-13. (Canceled)